BIS(1-TRIFLUOROMETHYL-2,2,2-TRIFLUOROETHOXY)TRIPHENYLPHOSPHORANE. A NEW ROUTE TO TRIFLUOROMETHYLATED HETEROCYCLES

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Toshio KUBOTA, Kazuhiro YAMAMOTO, and Tatsuo TANAKA

Department of Industrial Chemistry, Ibaraki Uiversity, Hitachi-shi,
Ibaraki 316

Bis(1-trifluoromethy1-2,2,2-trifluoroethoxy)triphenylphosphorane( $\underline{1}$ ) readily reacted with anilines to form N-hexafluoroisopropylated products( $\underline{2}$ ). By the dehydrofluorination of  $\underline{2}$ , 2-arylamino-1,1,3,3,3-pentafluoropropenes( $\underline{3}$ ) were obtained. Furthermore, when anilines with  $\varrho$ -functional group were used as nucleophile, benzo-1,4-dihetero-six membered rings that have CF $_3$  group at N-Q-position were obtained.

Phosphoranes with fluoroalkoxy group as ligand are known to be remarkable compounds as to the stereochemistry for phosphorus atom. However, little other utilities of these compounds in the synthesis of organofluorine compounds have been studied.  $^{2}$ 

On the other hand, heterocyclic compounds with trifluoromethyl group are attracting attention owing to their biological unique properties in these days.  $^{3}$ 

In our continuing study of phosphoranes,  $^{4-6)}$  we have found a new route to introduce the trifluoromethyl group into a desired position of heterocycles.

Bis(1-trifluoromethy1-2,2,2-trifluoroethoxy) triphenylphosphorane( $\underline{1}$ ) was readily reacted with anilines to produce the corresponding N-hexafluoroisopropylated derivertives( $\underline{2}$ ). This fact shows the remarkable contrast to analogous phosphorane, i.e., bis(2,2,2-trifluoroethoxy) triphenylphosphorane, which was inert to N-nucleophiles. Resulting N-hexafluoroisopropylanilines were dehydrofluorinated to form very interesting 2-arylamino-1,1,3,3,3-pentafluoropropenes( $\underline{3}$ ) by base, e.g., triethylamine, potassium tert-buthoxide, etc. However, it is impossible to obtain these compounds( $\underline{3}$ ) directly from the reaction of hexafluoropropene(HFP) with anilines, because HFP is attacked on the terminal CF $_2$  group by the nucleophiles.  $^{7}$ )

$$\begin{array}{c} \text{Ar-NH}_2 \xrightarrow{\text{Ph}_3 \text{P[OCH(CF}_3)_2]_2} \xrightarrow{\text{(1)}} & \text{Ar-NH-} \xleftarrow{\text{CF}_3} \xrightarrow{\text{Et}_3 \text{N}} & \text{Ar-NH-} \xleftarrow{\text{CF}_2} \\ & & \underline{2} & & \underline{3} & \text{CF}_3 \\ & & & \underline{2a} \, (\text{Ar = Ph}) & 46 \% & \underline{3a} & 76 \% \\ & & & \underline{2b} \, (\text{Ar=toly1}) & 58 \% & \underline{3b} & 81 \% \\ & & & & \underline{2b} \, (\text{HFP}) & & \underline{3} & & \\ & & & & \underline{3a} & 76 \% & \\ & & & & \underline{3b} & 81 \% \\ \end{array}$$

We also found that wider variety of anilines with o-functional group underwent the addition-elimination reaction to afford the trifluoromethylated heterocycles(4).

In conclusion, we believe that bis(1-trifluoromethy1-2,2,2-trifluoroethoxy)-triphenylphosphorane is useful intermediate to introduce the trifluoromethyl group into a desired position of heterocycles, and that this procedure provides a practical and convenient route to heterocycles.

In a typical procedure,  $\rho$ -phenylenediamine(1.80g, 10mmol) was added into a solution of  $\underline{1}$  in dichloromethane(40ml)-diethyl ether(10ml) which was prepared from triphenylphosphine dibromide(4.22g, 10mmol) and sodium 1-trifluoromethyl-2,2,2-trifluoroethoxide(3.80g, 20mmol) in situ. After 1h of stirring, precipitates(NaBr) were centrifuged from the reaction mixture, and the solvent was removed. then distillation gave N-hexafluoroisopropylated product( $\underline{2'a}$ , 1.51g) in a yield of 62%, b.p.84.0-85.0°C/7mmHg. H nmr(CDCl3):  $\Delta$ 4.40(NH), 4.50(NH2), 5.40(sep,  $J_{H-F}$ =6.1Hz, 1H), 6.20-6.75(4H). H nmr(CDCl3):  $\Delta$ 4.1.22(d, CF3) from CF3COOH as an ext. standard. MS: m/e 258(M<sup>+</sup>).

The mixture of 2'a(1.22g, 5mmol), triethylamine(1.50g, 15mmol) and tetrahydrofuran(30ml) was refluxed for 1h, and then worked-up as usual. 2-Fluoro-3-trifluoromethyl-1,4-dihydroquinoxaline( $\frac{4a}{4}$ , 0.80g) was obtained by column chromatography on silica gel using chloroform as eluent in 73% yield, m.p.89.5-91.0°C.  $^{1}_{H} \text{ nmr}(CDCl_{3}) : \text{$4.40(2H)$, 6.10-6.90(4H).} \quad ^{19}_{F} \text{ nmr}(CDCl_{3}) : \text{$6-15.9(d, J_{CF_{3}}-F^{=16.2Hz, CF_{3}}), -19.1(q, CF) from CF_{3}COOH as an ext. standard. MS : m/e 238(M+).}$ 

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